1,4-DIHYDROPYRIDINES AS INHIBITORS OF FREE-RADICAL REACTIONS

G. D. Tirzit and G. Ya. Dubur

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It is known [1, 2] that several 1,4-dihydropyridines (I) have hydrogen-donor properties. We have studied the behavior of I as inhibitors of free-radical reactions. We used the autooxidation of linetol [3] both in the pure state and as an emulsion in a phosphate buffer (pH 6.0) at 40°C for 6 h. The level of per-

oxidation was determined by means of iodometric titration (peroxide number) and also by reaction with 2-1 thiobarbituric acid (TBA) [4]. The inhibiting or antioxidant activity of I was expressed in percent of the reduction of the autooxidation. The results (the average values of 9 to 15 experiments) are presented in Table 1.

Using the electrochemiluminescence method, we also investigated I as inhibitors of free-radical reactions [5]. The results (average values of three to five experiments), expressed in percent decrease in the initial level of chemiluminescence (Jcl), are correlated in Table 2.

The results demonstrate that the investigated I inhibit the studied free-radical processes to the same extent as ionol and NADH₂.

TABLE 1. Antioxidant Activity of 1,4-Dihydropyridines I(R' = H) in Linetol

R	Linetolemulsion, 5·10 ⁻⁴ M (5·10 ⁻⁴ M of 1)		Linetol, 10 ⁻³ M of I				
	of I) without catalyst	FeSO ₄ catalysis, 5·10 ⁻⁴ M		atalyst	CuSO ₄ catalysis, 10 ⁻⁴ M		
	inhibition, %						
	with re- spect to TBA	with re- spect to TBA	with respect to peroxide No.	with re- spect to TBA	with respect to peroxide No.	with re- spect to TBA	
C₂H₅O CH₃ Ionol	57 84 57	57 56 77	72 53 100	41 79 100	60 18 97	41 16 40	

TABLE 2. Inhibition of the Electrochemiluminescence by 1,4-Dihydropyridines (I)

R	R'	Inhibition, % (7.7-10 ⁻⁷ M of I)	Conen. of I that reduces J _{c1} by 50%, M
C ₂ H ₅ O CH ₃ CH ₃ CH ₃ NaDH ₂ Na-Salt	COONa COONa H CH ₃	20 14 15 14 18	4 · 10 ⁻⁵ 2 · 10 ⁻⁵ 8 · 10 ⁻⁶

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We feel that a new class of antioxidants and free-radical-reaction inhibitors - 1,4-dihydropyridines - has been discovered.

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